# ON SOME LIPIDIC COMPONENTS FROM THE ROOT OF Petasites hybridus (L.) G. M. SCH.\*

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In the chloroform extract from the root of *Petasites hybridus* (L.) G. M. SCH. there were detected by means of chromatographic and spectrophotometric methods in addition to compounds of the terpenoid nature the n-alkanes, esters of sterols (mainly of  $\beta$ -sitosterol) with higher aliphatic acids (mainly the C<sub>16:0</sub>, C<sub>18:2</sub>, and C<sub>18:3</sub> acids), and triglycerides containing the same acids.

In connection with investigations on the constituents<sup>1-3</sup> of *Petasites hybridus* (L.) G. M. SCH. (Syn.: *P. officinalis* MOENCH.) we have been interested in some so far poorly examined lipidic components. In the present paper, we wish to report on the analysis of the chloroform extract of roots.

#### **EXPERIMENTAL**

*Material.* Slices of the root were dried at room temperature, powdered (2.50 kg), the powder placed into a column, and washed at room temperature with chloroform (121). Evaporation of the chloroform extract yielded 62 g of the residue. A portion (20 g) was dissolved in chloroform (100 ml) and adsorbed to 50 g of deactivated silica gel. The solvent was removed and the residue chromatographed on a column ( $5.6 \times 68$  cm) of the Pitra<sup>4</sup> silica gel (particle size, 0.10-0.25 mm; 800 g) previously deactivated by the addition of water (12.5%). Light petroleum (b.p. 45-65°C) containing increasing amounts of ether (up to 20%) was used as eluant. The fractions were monitored by thin-layer chromatography on silica gel as well as by gas chromatography.

Methods. Gas chromatography was performed on a Perkin-Elmer F 11 apparatus with flame ionisation detectors and a dual system of glass columns ( $0.4 \times 150$  cm) packed with 3% SE-30 G.C. Grade on Gas-Chrom Q (80—100 mesh) or 20% diethylene glycol succinate on HMDS-Chromosorb W (80—100 mesh). Gas chromatography-mass spectrometry was performed on a combination of PYE Series 104 Chromatograph Model 64 and A.E.J. MS 902 apparatus with the use of the Watson-Biemann separator. Ionic source temperature, 150—200°C (according to the substance type); electron energy, 70 eV. Infrared spectroscopy was carried out in a UR-20 apparatus in 0.01 cm cells. Concentration of samples, 6% in tetrachloromethane.

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### RESULTS

n-Alkanes (0.11% of the extract residue) form a homologous series  $(C_{15}--C_{37})$  with dominating odd members. The  $C_{23}$  (7.9%),  $C_{25}$  (15.2%),  $C_{27}$  (30.9%),  $C_{29}$  (23.5%), and  $C_{31}$  (7.8%) homologues were the most abundant. It may be seen from the chromatographic record that the n-alkanes are as usually accompanied by a small amount of monomethylalkanes<sup>5</sup>.

Terpenoids (4.0% of the extract residue). A systematic identification of sesquiterpenoid hydrocarbons has been performed earlier<sup>6,7</sup>. Nevertheless, some additional components of column chromatography fractions were completely or partly identified by means of gas chromatography-mass spectrometry. Thus, the presence of M == 152 (base peak M - (26 + 15)) and M = 162 (main fragments 95, M - 15, M - 29) substances was recorded. There was also present a sesquiterpenic alcohol (M = 220; main fragments 41, 91, 81) absorbing in the IR region at 3620, 3440 and 1040 cm<sup>-1</sup>. Moreover, the sesquiterpenic hydrocarbon  $\beta$ -santalene<sup>8</sup> (M = 204: main fragments 94, 41, 122) was newly identified. The subsequent fraction obtained from the chromatographic column (0.63% of the extract residue) mainly contains sesquiterpenic lactones<sup>2,3,6</sup> with IR absorption at 1770 and  $1193 \text{ cm}^{-1}$  (lactone group), at 3620 and 3440 cm<sup>-1</sup> (-OH), at 1040 cm<sup>-1</sup> (-O-), and at 1594 cm<sup>-1</sup> (>C=C<). The most abundant components showed the following values: M = 218 (main fragments  $103 \ge 123$ , 218; 23.6%), M = 234 (main fragments M - 2, 123, 41, 109, 110; 49.3%), M = 234 (main fragments M - 2, 123, 109, 41, 110; 9.9%), and M = 232 (main fragments 109, 41, 91, 123, 232, 161; 9.0%).

*Esters.* The subsequent fraction (6.9% of the extract residue) exhibited IR absorption bands of the ester grouping (1732, 1249 and 1178 cm<sup>-1</sup>) and of the double

Acid	A (%)	B (%)	Acid	A (%)	B (%)
12:0	× ···	a	17:1	0.3	0.2
14:0	0.5	0.6	18:0	1.0	1.3
15:0	0.4	1.8	18:1	2.0	3.7
?	0.8		18:2	64.3	54.8
16:0	12.9	16.1	18:3	17.2	18.5
16:1	0.6	3.0	20:0	а	a
17:0	·	а	22:0	a	

 TABLE I

 Fatty Acids Obtained from Esters (A) and Glycerides (B)

<sup>a</sup> Trace amount.

bond (1642 cm<sup>-1</sup>). After the transesterification<sup>9</sup>, there was isolated a mixture of methyl esters of higher fatty acids (Table I) and a mixture of sterols. By means of mass spectrometry of trimethylsilylated hydroxy derivatives<sup>10-12</sup>, the C<sub>16:1</sub> acid was identified as 9-hexadecenoic acid, the C<sub>18:1</sub> acid as 9-octadecenoic acid (oleic acid), and the C<sub>18:2</sub> acid as 9,12-octadecadienoic acid (linoleic acid). This method failed in determination of double bond positions in the C<sub>18:3</sub>; the retention times in gas chromatography corresponded to those of authentic linolenic acid. The sterols were identified by gas chromatography and mass spectrometry<sup>13</sup> as campesterol (9·3%), stigmasterol (7·9%), and β-sitosterol<sup>6</sup> (72·8%); identification of the remaining three sterols (6·1, 2·9, and 1·0%) failed by this method.

Triglycerides. The subsequent chromatographic fraction (5.9% of the extract residue) exhibited the IR absorption at 1745 and 1162 cm<sup>-1</sup> (ester group) and at 3015 cm<sup>-1</sup> ( $\geq C = C$ ). On thin-layer chromatography (silica gel), the  $R_F$  value was identical with that of the synthetic triglyceride trimyristin. After the transesterification<sup>9</sup>, there was isolated and identified by gas chromatography a mixture of higher fatty acids in the form of methyl esters (Table I); glycerol was identified in the form of the trimethylsilyl derivative<sup>14</sup>. As it may be inferred from Table I, composition of fatty acids in glycerides is very similar to that of acids bound to sterols.

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